

Synthesis of the C8–C20 and C21–C30 segments of pectenotoxin 2

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Abstract—In this study, we synthesized the C8–C20 and C21–C30 segments of the diarrhetic shellfish toxin pectenotoxin 2. The C8–C20 segment was assembled from a phosphonate corresponding to the C8–C15 segment (prepared from L-malic acid in 19 steps) and an aldehyde corresponding to the C16–C20 segment (synthesized from 3-methyl-3-butenol in nine steps) by a twelve-step process including the Horner–Wadsworth–Emmons reaction, regio- and stereoselective reduction of the resulting enone, diastereoselective epoxidation, and 5-*exo* epoxide cleavage forming the C-ring. The C21–C30 segment was constructed in 13 steps from (*S*)-glycidol via a route involving E-ring formation by 5-*exo* epoxide cleavage and stereoselective methylation at C27 by the Evans method. © 2007 Elsevier Ltd. All rights reserved.

The pectenotoxin (PTX) family of diarrhetic shellfish toxins, which were isolated from toxin infected scallop *Patinopecten yessoensis* and the dinoflagellate *Dinophysis fortii* by Yasumoto,¹ has an unusual thirty-four-membered macrolactone that includes a spirocyclic acetal AB-ring, a six-membered cyclic hemiacetal G-ring, a bicyclic acetal D-ring, and three oxolanes (C, E, and F-rings). While some PTXs show potent hepatotoxicity in mice,² recent studies have reported that PTX2 (**1**) (Fig. 1) also exhibits strong cytotoxicity against cancer cells³ and actin-depolymerizing activity.⁴ These

remarkable structural and biological features of PTXs have attracted the attention of synthetic organic chemists.^{5–7} As part of our goal toward total synthesis of PTXs, we describe herein the synthesis of the C8–C20 and C21–C30 segments (**2** and **3**, respectively; Scheme 1) of PTX2 (**1**).

From the retrosynthetic perspective (Scheme 1), guided by our previous synthesis of the C8–C18 segment of PTX2,^{7b} the C8–C20 synthetic segment (**2**) was envisioned to arise via a 5-*exo* epoxide cleavage reaction of **4**, which would be assembled from phosphonate **5** and aldehyde **6** through a route involving the Horner–Wadsworth–Emmons reaction, regio- and stereoselective reduction of the ketone at C14, and diastereoselective epoxidation. Although we previously synthesized the C8–C15 segment, which is structurally the same as **5** except for the protective group of the oxygen at C11, it required a lengthy 26 step process from 3-butenol.^{7b} Therefore, we intended to prepare **5** by an alternative shorter route including formation of enone **9** from **10**, diastereoselective reduction of **9** affording **8**, diastereoselective epoxidation of **8** followed by protection giving **7**, and construction of the β-keto-phosphonate part of **5**. Aldehyde **6**, having a quaternary asymmetric center at C18, would be derived from epoxy alcohol **11** exploiting regioselective reductive cleavage followed by oxidation. We also undertook the synthesis of the C21–C30 segment (**3**) from **12** through the Evans alkylation⁸ to make

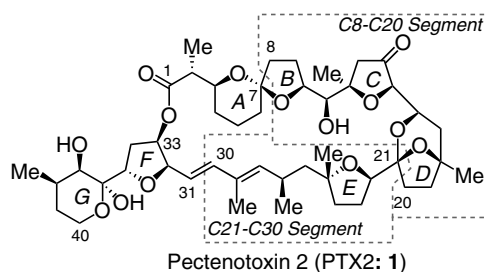
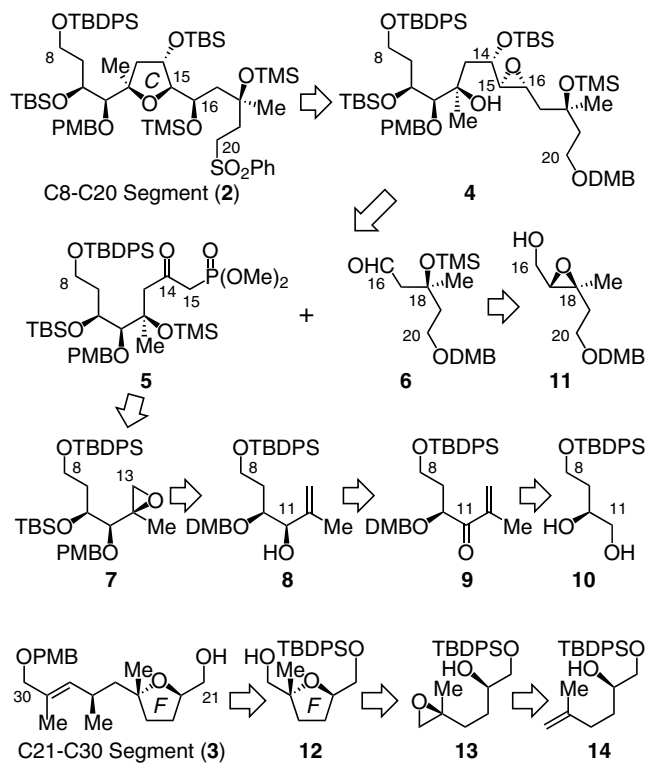


Figure 1.

Keywords: Pectenotoxin 2; Natural product synthesis; Polyether macrolide.

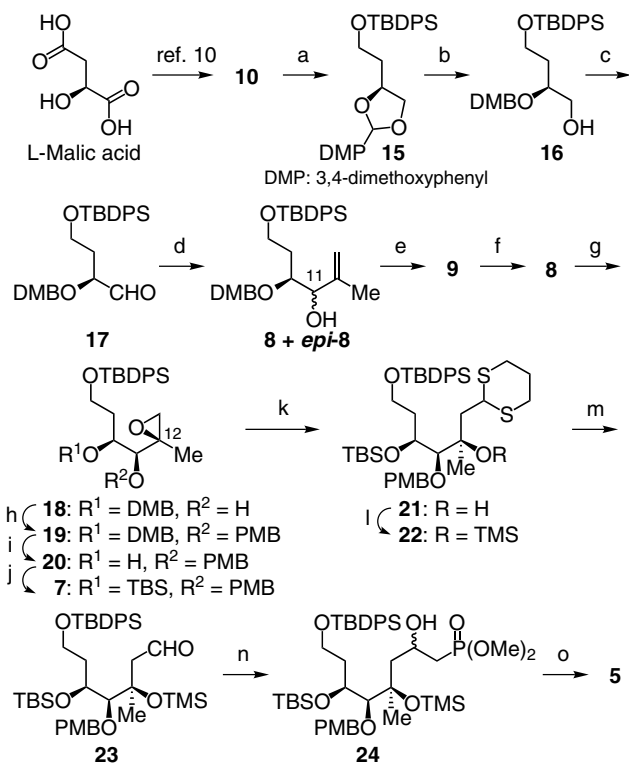
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Scheme 1.

the stereocenter at C27 and the Wittig reaction to form the trisubstituted double bond at C28 (Scheme 1). In turn, the E-ring of 12 would arise via diastereoselective epoxidation of bishomoallyl alcohol 14 followed by the 5-*exo* epoxide cleavage reaction of the resulting epoxide 13.⁹

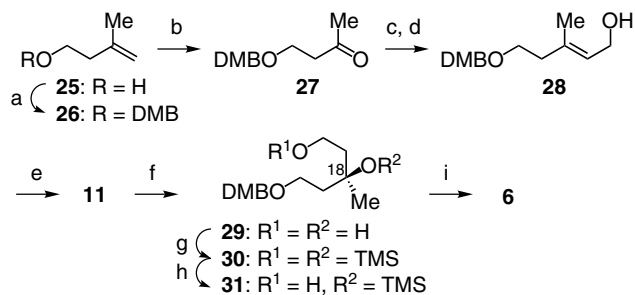
Scheme 2 illustrates synthesis of phosphonate 5, starting from known diol 10 prepared in four steps from L-malic acid.¹⁰ Diol 10 was converted to 3,4-dimethoxy benzyl (DMB) ether 16 through acetalization (47%: 79% based on recovery of 10) followed by reductive cleavage with DIBALH (71%). Swern oxidation¹¹ of 16 and the subsequent reaction with 2-propenylmagnesium bromide gave a 1:1 mixture of 8 and its diastereomer (*epi*-8) (68% from 12). To establish stereochemistry at C11, an oxidation/reduction process was applied. Swern oxidation¹¹ of the mixture of 8 and *epi*-8 (90%) and reduction of the resulting 9 with Zn(BH₄)₂ afforded 8 as a sole product (71%).^{12,13} Alcohol 8 was epoxidized with TBHP in the presence of VO(acac)₃ to produce 18 as an almost single diastereomer (67%),^{14,15} which was then converted to 7 through a three-step process [protection with PMBBR (83%), removal of DMB, and protection with TBSOTf (71% from 19)]. Following Nakata's procedure, epoxide 7 was reacted with 1,3-dithiane to give 21 in good yield (72%).¹⁶ After TMS-protection of 21, the dithiane group was hydrolyzed with Hg(ClO₄)₂ in the presence of 2,6-di-*tert*-butylpyridine to produce 23 (72%),^{17,18} which was reacted with lithiated dimethyl methylphosphonate to afford 24 (97%). Finally, alcohol 24 was oxidized with Dess–Martin periodinane (DMPI) to produce 5 (100%).¹⁹ Phosphonate 5 was thus synthesized from L-malic acid in 19 steps.



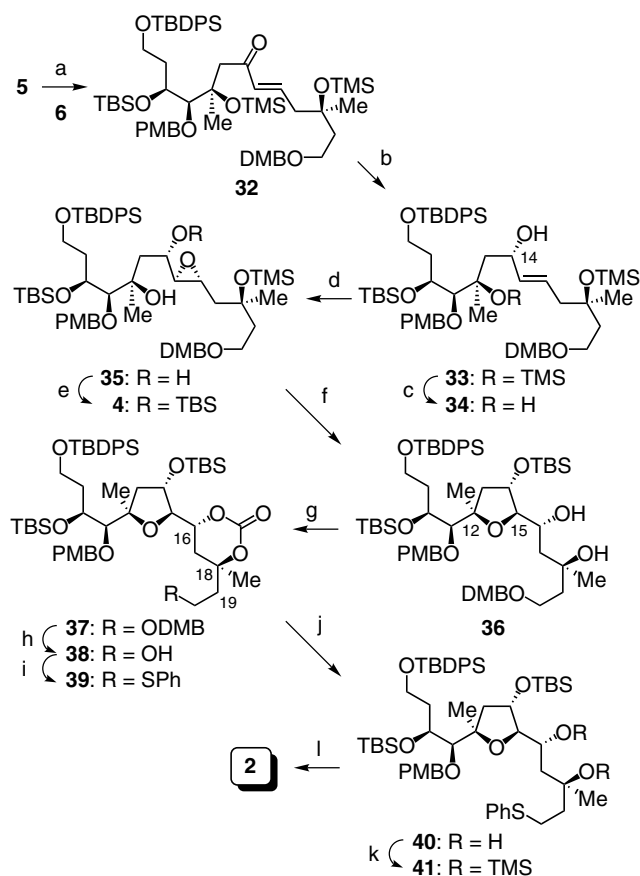
Scheme 2. Reagents and conditions: (a) 3,4-dimethoxybenzaldehyde, CSA (cat), toluene, reflux, 5 h, 15: 47%, recovered 10: 41%; (b) DIBALH, CH₂Cl₂, -78 to 0 °C, 30 min, 71%; (c) (COCl)₂, DMSO, CH₂Cl₂, -78 °C, 10 min, then Et₃N, -78 to 0 °C, 15 min; (d) 2-propenylmagnesium bromide, THF, -78 °C, 10 min, 68% from 16; (e) (COCl)₂, DMSO, CH₂Cl₂, -78 °C, 10 min, then Et₃N, -78 to 0 °C, 15 min, 90%; (f) Zn(BH₄)₂, Et₂O, -30 °C, 20 min, 71%; (g) VO(acac)₃, TBHP, CH₂Cl₂, 0 °C, 3.5 h, 67%; (h) NaH, PMBBR, Bu₄NI, THF, 23 °C, 2 h, 83%; (i) DDQ, CH₂Cl₂-pH 7 buffer (10:1), 26 °C, 3.5 h; (j) TBSOTf, 2,6-lutidine, CH₂Cl₂, 0 °C, 1 h, 71% from 19; (k) 1,3-dithiane, *t*-BuLi, Bu₂Mg, THF, -20 °C, 30 min, then 7, 23 °C, 1 h, 72%; (l) TMSOTf, 2,6-lutidine, DMF, 0 °C, 70 min, 87%; (m) Hg(ClO₄)₂, 2,6-di-*tert*-butylpyridine, THF-H₂O (5:1), 25 °C, 15 min, 72%; (n) (MeO)₂P(O)CH₃, BuLi, THF, -78 °C, 30 min, then 23, -78 °C, 2 h, 97%; (o) DMPI, NaHCO₃, CH₂Cl₂, 23 °C, 5 min, 100%.

Aldehyde 6 was synthesized from 3-methyl-3-butenol (25) (Scheme 3). Protection of 25 as a DMB ether (64%) followed by a one-pot dihydroxylation/diol-cleavage process afforded 27 (68%), which was converted to 28 stereoselectively via the Horner–Wadsworth–Emmons reaction (99%: *E/Z* = ~2.7:1) and DIBALH reduction (73% after separation). Katsuki–Sharpless asymmetric epoxidation²⁰ of 28 with (+)-diisopropyl tartrate gave 11 (97%) in good optical yield (95% ee).²¹ Epoxide 11 was regioselectively cleaved with LiAlH₄ to produce 29 (100%),²² which was transformed to 31 by a stepwise protection/deprotection process (95% and 89%, respectively). Alcohol 31 was oxidized with TPAP and NMO to aldehyde 6.²³

Scheme 4 shows the synthesis of C8–C20 segment 2. The coupling reaction of 5 with 6 and the subsequent C-ring formation was performed following our previous procedure.^{7b} Phosphonate 5 was coupled with 6 by the Horner–Wadsworth–Emmons reaction to afford 32 (85%), which was stereoselectively reduced to 33 under Luche



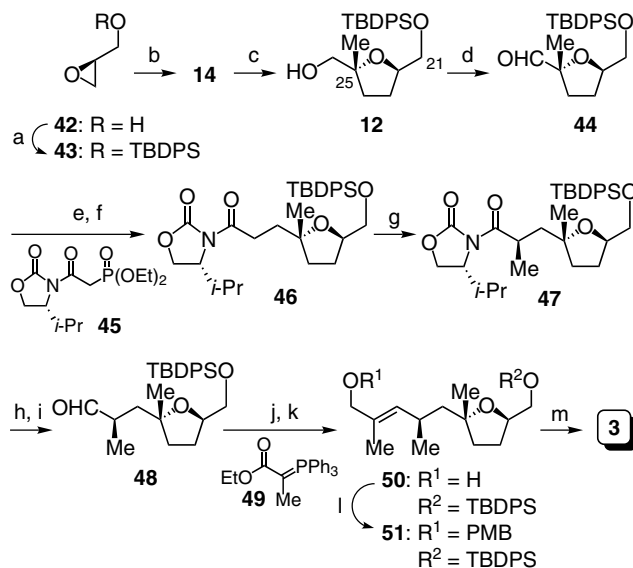
Scheme 3. Reagents and conditions: (a) NaH, 3,4-dimethoxybenzyl chloride, Bu₄NI, THF, 25 °C, 24 h, 64%; (b) OsO₄, NMO, 1,4-dioxane–pH 7 buffer (3:1), 23 °C, 1 h, then NaIO₄, 3 h, 68%; (c) (EtO)₂(O)PCH₂CO₂Et, NaH, THF, 23 °C, 12.5 h, 99% (*E/Z* = ~2.7:1); (d) DIBALH, CH₂Cl₂, –78 °C, 40 min, then separation, 73%; (e) (+)-diisopropyl tartrate, Ti(O*i*-Pr)₄, TBHP, MS4A, CH₂Cl₂, –30 °C, 12 h, 97% (95% ee); (f) LiAlH₄, THF, 24 °C, 1 h, 100%; (g) TMSOTf, 2,6-lutidine, CH₂Cl₂, –40 °C, 45 min, 95%; (h) K₂CO₃, MeOH, 23 °C, 90 s, 89%; (i) TPAP, NMO, MS4A, CH₂Cl₂, 0 °C, 30 min, 87%.



Scheme 4. Reagents and conditions. (a) NaH, benzene–THF (2:5), 0 °C, 5 min, then 6, –78 °C, 8 h, 85%; (b) NaBH₄, CeCl₃·H₂O, EtOH, –20 °C, 3 h, 98%; (c) K₂CO₃, MeOH, 24 °C, 2 h, 93%; (d) *m*-CPBA, NaHCO₃, CH₂Cl₂, 0 °C, 2 h, 95% (dr = 3:1); (e) TBSOTf, 2,6-lutidine, CH₂Cl₂, –20 °C, 1 h, then separation, 66%; (f) CSA (cat.), CH₂Cl₂, 0 °C, 4 h, 77%; (g) (CCl₃O)₂CO, pyridine, CH₂Cl₂, 0 °C, 2 h, 95%; (h) NaBH₄, THF–H₂O (3:1), 23 °C, 22 h, 100%; (i) (COCl)₂, DMSO, CH₂Cl₂, –78 to 0 °C, 30 min, 97%; (j) 49 (excess), benzene, reflux, 56 h; (k) DIBALH, CH₂Cl₂, –78 °C, 25 min, 92% from 48; (l) NaH, PMBBR, Bu₄NI, THF, 23 °C, 16 h, 93%; (m) *m*-CPBA, NaHCO₃, CH₂Cl₂, 25 °C, 1 h, 100%.

conditions (98%).^{24,25} Selective detachment of the TMS group at C12-oxygene (93%) followed by hydroxy-directed epoxidation with *m*-CPBA gave a 3:1 mixture of epoxide 35 and its diastereomer (95%).²⁶ After selective protection with TBSOTf, epoxide 4 was isolated with a 66% yield. Intramolecular 5-*exo* cyclization of 4 with a catalytic amount of CSA led to 36 (77%).²⁷ Thus, the C-ring was successfully constructed. Diol 36 was then transformed with triphosgene to carbonate 37 (95%), of which the DMB group was removed with TMSOTf/2,6-lutidine to give 38 (95%).²⁸ Installation of a phenylthio group to 38 with *N*-(phenylthio)phthalimide/Bu₃P to afford 39 (94%)²⁹ followed by a two-step process including removal of the carbonate of 39 with LiAlH₄ (100%) and protection of diol 40 with TMSOTf produced 41 in good yield (93%). Finally, sulfide 41 was oxidized with *m*-CPBA (100%), thereby completing the synthesis of the C8–C20 segment (2)³⁰ of PTX2 (1).

C21–C30 segment 3 was constructed from (*S*)-glycidol (42) (Scheme 5). Protection of 42 with TBDPSCI (100%) and the subsequent reaction with 2-methyl-3-propenylmagnesium chloride afforded 14 (100%). When bishomoallyl alcohol 14 was subjected to epoxidation with TBHP/VO(acac)₂ followed by treatment with CSA, the desired oxolane 12 was produced predominantly (57%) along with its diastereomer (*epi*-12: 18%) and recovered 14 (11%).^{9,31} Swern oxidation of 12 afforded 44 (96%),¹¹ which was converted to 46 by the



Scheme 5. Reagents and conditions: (a) TBDPSCI, imidazole, DMF, 25 °C, 1 h, 100%; (b) 3-chloro-2-methylpropene, Mg, THF, 24 °C, 1 h, then 43, –40 °C, 5 h, 100%; (c) VO(acac)₂ (cat.), TBHP, benzene, 55 °C, 21 h, then CSA (cat.), 21 °C, 2 h, 12: 57%, *epi*-12: 18%, recovered 14: 11%; (d) (COCl)₂, DMSO, CH₂Cl₂, –78 °C, 35 min, then Et₃N, –78 to 0 °C, 75 min, 96%; (e) 45, NaH, THF, 23 °C, 30 min, then 44, 23 °C, 1.5 h, 93%; (f) H₂, Pd/C, EtOH, 23 °C, 16 h, 100%; (g) NaHMDS, THF, –78 °C, 1 h, then MeI, –78 °C, 4 h, 93%; (h) NaBH₄, THF–H₂O (3:1), 23 °C, 22 h, 100%; (i) (COCl)₂, DMSO, CH₂Cl₂, –78 to –50 °C, 15 min, then Et₃N, –50 to 0 °C, 30 min, 97%; (j) 49 (excess), benzene, reflux, 56 h; (k) DIBALH, CH₂Cl₂, –78 °C, 25 min, 92% from 48; (l) NaH, PMBBR, Bu₄NI, THF, 23 °C, 16 h, 93%; (m) Bu₄NF, THF, 23 °C, 1.5 h, 100%.

Horner–Wadsworth–Emmons reaction with **45** (93%)³² and the subsequent hydrogenation (100%). Stereoselective methylation of **46** by the Evans method to give **47** (93%), reductive detachment of the chiral auxiliary (100%), and then Swern oxidation provided **48** (97%).¹¹ Aldehyde **48** was reacted with Wittig reagent **49**, and the resulting unsaturated ester was reduced with DIBALH to afford **50** stereoselectively (92% from **48**). The final protection/deprotection process transformed **50** to **3**^{33,34} (96% for two steps). Thus, the synthesis of the C21–C30 segment (**3**) of PTX2 (**1**) was accomplished.

In conclusion, we successfully synthesized the C8–C20 and C21–C30 segments (**2** and **3**, respectively) of PTX2 (**1**). C8–C20 segment **2** was assembled from phosphonate **5** (prepared from L-malic acid in nineteen steps) and aldehyde **6** (synthesized from 3-methyl-3-butenol in nine steps) by a twelve-step process including the Horner–Wadsworth–Emmons reaction, regio- and stereoselective reduction of the resulting enone, diastereoselective epoxidation, and 5-*exo* epoxide cleavage forming the C-ring. C21–C30 segment **3** was constructed in thirteen steps from (*S*)-glycidol via a route involving E-ring formation by 5-*exo* epoxide cleavage, stereoselective methylation at C27, and formation of the trisubstituted olefin at C28. All newly generated stereocenters of **2** and **3** were properly confirmed by the NMR or $[\alpha]_D$ analysis of the intermediates or derivatives from the intermediates (Figs. 2–4).^{13,15,22,25,27,31,33} Further efforts toward the total synthesis of PTX2 are currently underway in our laboratory.

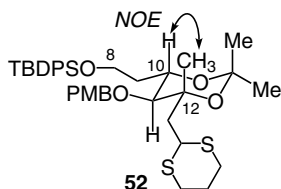


Figure 2.

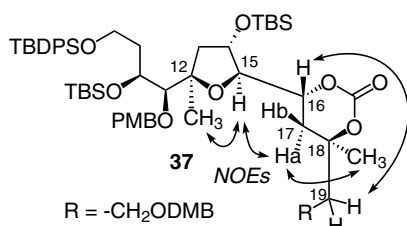


Figure 3.

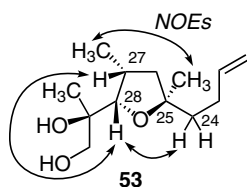


Figure 4.

Acknowledgments

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References and notes

- (a) Yasumoto, T.; Murata, M.; Oshima, Y.; Sano, M.; Matsumoto, G. K.; Clardy, J. *Tetrahedron* **1985**, *41*, 1019; (b) Sasaki, K.; Wright, J. L. C.; Yasumoto, T. *J. Org. Chem.* **1998**, *63*, 2475; (c) Sasaki, K.; Satake, M.; Yasumoto, T. *Biosci. Biotech. Biochem.* **1997**, *61*, 1783.
- (a) Ishige, M.; Sato, N.; Yasumoto, T. *Rep. Hokkaido Inst. Public Health* **1988**, *38*, 15; (b) Terao, K.; Ito, E.; Yanagi, T.; Yasumoto, T. *Toxicon* **1986**, *24*, 1141.
- Jung, J. H.; Sim, C. J.; Lee, C.-O. *J. Nat. Prod.* **1995**, *58*, 1722.
- (a) Zhou, Z.-H.; Komiyama, M.; Terao, K.; Shimada, Y. *Nat. Toxins* **1994**, *2*, 132; (b) Hori, M.; Matsuura, Y.; Yoshimoto, R.; Ozaki, H.; Yasumoto, T.; Karaki, H. *Folia Pharmacol. Jpn.* **1999**, *114*, 225; (c) Spector, I.; Braet, F.; Choquet, N.; Bubbs, M. R. *Microsc. Res. Technol.* **1999**, *47*, 18; (d) Leira, F.; Cabado, A. G.; Vieytes, M. R.; Roman, Y.; Alfonso, A.; Botana, L. M.; Yasumoto, T.; Malaguti, C.; Rossini, G. P. *Biochem. Pharmacol.* **2002**, *63*, 1979.
- The total synthesis of pectenotoxins-4 and -8, see: (a) Evans, D. A.; Rajapakse, H. A.; Stenkamp, D. *Angew. Chem., Int. Ed.* **2002**, *41*, 4569; (b) Evans, D. A.; Rajapakse, H. A.; Chiu, A.; Stenkamp, D. *Angew. Chem., Int. Ed.* **2002**, *41*, 4573.
- Other synthetic studies, see: (a) Micalizino, G. C.; Roush, W. R. *Org. Lett.* **2001**, *3*, 1949; (b) Paquette, L. A.; Peng, X.; Bonder, D. *Org. Lett.* **2002**, *4*, 937; (c) Pihko, P. M.; Aho, J. E. *Org. Lett.* **2004**, *6*, 3849; (d) Peng, X.; Bonder, D.; Paquette, L. A. *Tetrahedron* **2004**, *60*, 9589; (e) Bonder, D.; Liu, J.; Muller, T.; Paquette, L. A. *Org. Lett.* **2005**, *7*, 1813; (f) Halim, R.; Brimble, M. A.; Merten, J. *Org. Lett.* **2005**, *7*, 2659; (g) Halim, R.; Brimble, M. A.; Merten, J. *Org. Biomol. Chem.* **2006**, *4*, 1387; (h) Vellucci, D.; Rychnovsky, S. D. *Org. Lett.* **2007**, *9*, 711; (i) Lotesta, S. D.; Hou, Y.; Williams, L. J. *Org. Lett.* **2007**, *9*, 369; (j) O'Connor, P. D.; Knight, C. K.; Friedrich, D.; Peng, X.; Paquette, L. A. *J. Org. Chem.* **2007**, *72*, 1747; A review, see: (k) Halim, R.; Brimble, M. A. *Org. Biomol. Chem.* **2006**, *4*, 4048.
- (a) Amano, S.; Fujiwara, K.; Murai, A. *Synlett* **1997**, 1300; (b) Awakura, D.; Fujiwara, K.; Murai, A. *Synlett* **2000**, 1733; (c) Fujiwara, K.; Kobayashi, M.; Yamamoto, F.; Aki, Y.; Kawamura, M.; Awakura, D.; Amano, S.; Okano, A.; Murai, A.; Kawai, H.; Suzuki, T. *Tetrahedron Lett.* **2005**, *46*, 5067.
- Evans, D. A.; Ennis, M. D.; Mathre, D. J. *J. Am. Chem. Soc.* **1982**, *104*, 1737.
- Fukuyama, T.; Varanesic, B.; Negri, D. P.; Kishi, Y. *Tetrahedron Lett.* **1978**, *19*, 2741.
- (a) Hayashi, H.; Nakanishi, K.; Brandon, C.; Marmur, J. *J. Am. Chem. Soc.* **1973**, *95*, 8749; (b) Mori, Y.; Takeuchi, A.; Kageyama, H.; Suzuki, M. *Tetrahedron Lett.* **1988**, *29*, 5423; (c) Clive, D. L. J.; Murthy, K. S. K.; Wee, A. G. H.; Prasad, J. S.; da Silva, G. V. J.; Majewski, M.; Anderson,

- P. C.; Evans, C. F.; Haugen, R. D.; Heerze, L. D.; Barrie, J. R. *J. Am. Chem. Soc.* **1990**, *112*, 3018.
11. Mancuso, A. J.; Huang, S.-L.; Swern, D. *J. Org. Chem.* **1978**, *43*, 2048.
12. Nakata, T.; Oishi, T. *Tetrahedron Lett.* **1980**, *21*, 1641.
13. The absolute stereochemistry at C11 of **8** was determined by new Mosher's method: Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, *113*, 4092.
14. (a) Rossiter, B. E.; Verhoeven, T. R.; Sharpless, K. B. *Tetrahedron Lett.* **1979**, *20*, 4733; (b) Tomioka, H.; Suzuki, T.; Nozaki, H.; Oshima, K. *Tetrahedron Lett.* **1982**, *23*, 3387.
15. The configuration at C12 of **18** was determined by the presence of NOE between H10 and C12–CH₃ of isopropylidene acetal **52** derived from **18** via **21** (Fig. 2).
16. Ide, M.; Nakata, M. *Bull. Chem. Soc. Jpn.* **1999**, *72*, 2491.
17. Fujita, E.; Nagao, Y.; Kaneko, K. *Chem. Pharm. Bull.* **1978**, *26*, 3743.
18. The use of 2,6-di-*tert*-butylpyridine was essential to obtain reproducible result.
19. (a) Dess, D. B.; Martin, J. C. *J. Org. Chem.* **1983**, *48*, 4155; (b) Dess, D. B.; Martin, J. C. *J. Am. Chem. Soc.* **1991**, *113*, 7277.
20. Katsuki, T.; Sharpless, K. B. *J. Am. Chem. Soc.* **1980**, *102*, 5974.
21. The optical yield of **11** was determined by HPLC using a chiral column [Daicel Chiralcel AD, eluent: hexane/2-propanol (9:1)].
22. The absolute stereochemistry at C18 of **29** was confirmed by the conversion of **29** to known (3*S*)-5-(*tert*-butyldiphenylsilyloxy)-3-methyl-3,5-pentandiol via a two-step process [(i) TBDPSCl, imidazole; (ii) H₂, Pd/C]. The $[\alpha]_D^{20}$ value of the synthetic compound was in good agreement with the reported value $\{[\alpha]_D^{19} -7.7$ (CHCl₃, *c* 0.975); lit. $[\alpha]_D^{20} -6.61$ (CHCl₃, *c* 0.5): Krohn, K.; Meyer, A. *Liebigs Ann. Chem.* **1994**, 167}.
23. Griffith, W. P.; Ley, S. V. *Aldrichim. Acta* **1990**, *23*, 13.
24. Luche, J.-L.; Gemal, A. L. *J. Am. Chem. Soc.* **1979**, *101*, 5848.
25. The absolute stereochemistry at C14 of **33** was determined by new Mosher's method. See Ref. 13.
26. The presence of the TMS group at C12-oxygen reduced stereoselectivity of the epoxidation step.
27. The configurations at C15 and C16 of oxolane **36** were confirmed by the presence of NOEs between H15 and C12–CH₃, between H15 and H17a, between H17a and C18–CH₃, and between H16 and H19 in **37** (Fig. 3).
28. We found that a primary alkyl DMB ether was readily removed on treatment with TMSOTf/2,6-lutidine. This phenomenon was helpful for selective detachment of the DMB group of **37** in the presence of the PMB ether at C11. On the other hand, detachment of a DMB group from **29** during TMS ether formation was avoided by lowering the reaction temperature (Scheme 3). cf. Oriyama, T.; Yatabe, K.; Kawada, Y.; Koga, G. *Synlett* **1995**, 45.
29. Walker, K. A. M. *Tetrahedron Lett.* **1977**, *18*, 4475.
30. Selected spectral data of **2**: A colorless oil; $[\alpha]_D^{22} +13.2$ (*c* 1.43, CHCl₃); ¹H NMR (300 MHz, C₆D₆, C₆H₅D₅ as 7.15 ppm) δ 0.04 (3H, s), 0.08 (3H, s), 0.13 (3H, s), 0.14 (9H, s), 0.21 (3H, s), 0.28 (9H, s), 0.95 (9H, s), 0.97 (9H, s), 1.21 (9H, s), 1.29 (3H, s), 1.50–1.60 (1H, m), 1.56 (3H, s), 1.70 (1H, dd, *J* = 3.0, 12.7 Hz), 1.98–2.06 (2H, m), 2.12–2.30 (2H, m), 2.29 (1H, dd, *J* = 7.6, 12.7 Hz), 3.31–3.40 (1H, m), 3.36 (3H, s), 3.51 (1H, dt, *J* = 5.8, 12.7 Hz), 3.70 (1H, s), 3.86–3.99 (2H, m), 4.09–4.12 (2H, m), 4.21 (1H, dd, *J* = 2.6, 5.3 Hz), 4.50 (1H, br t, *J* = 6.0 Hz), 4.74 (1H, d, *J* = 11.2 Hz), 5.02 (1H, d, *J* = 11.2 Hz), 6.86–6.94 (5H, m), 7.04–7.29 (6H, m), 7.38 (2H, d, *J* = 8.8 Hz), 7.77–7.83 (4H, m), 7.87–7.91 (2H, m); ¹³C NMR (75 Hz, C₆D₆, ¹³C¹²C₅D₆ as 128.0 ppm) δ -4.7 (CH₃), -4.6 (CH₃), -4.3 (CH₃), -3.3 (CH₃), 1.0 (CH₃ × 3), 2.6 (CH₃ × 3), 18.0 (C), 18.3 (C), 19.4 (C), 23.2 (CH₃), 25.9 (CH₃ × 3), 26.2 (CH₃ × 3), 27.2 (CH₃ × 3), 28.3 (CH₃), 36.0 (CH₂), 36.5 (CH₂), 42.9 (CH₂), 45.0 (CH₂), 52.8 (CH₂), 54.8 (CH₃), 61.3 (CH₂), 70.3 (CH), 70.9 (CH), 74.3 (CH), 74.5 (CH₂), 75.1 (C), 85.2 (C), 88.5 (CH), 89.4 (CH), 114.1 (CH × 2), 128.05 (CH × 2), 128.07 (CH × 2), 128.5 (CH × 2), 129.1 (CH × 2), 129.3 (CH × 2), 130.0 (CH × 2), 131.8 (C), 132.9 (CH), 134.3 (C × 2), 135.97 (CH × 2), 136.03 (CH × 2), 141.1 (C), 159.7 (C); IR (film) ν_{\max} 3070, 2957, 2935, 2895, 2850, 1612, 1587, 1513, 1472, 1460, 1446, 1428, 1390, 1375, 1361, 1310, 1302, 1251, 1087, 940, 910, 904, 839, 775, 745, 702, 695 cm⁻¹; HR-FDMS calcd for C₆₃H₁₀₄O₁₀Si₅S [M⁺]: 1192.6196; found, 1192.6213.
31. The stereochemistry at C25 of **12** was determined by the presence of NOE between H21 and C25–CH₃.
32. Koch, S. S. C.; Chamberlin, A. R. *J. Org. Chem.* **1993**, *58*, 2725.
33. The stereochemistry at C27 of **3** was confirmed by the presence of NOEs between C25–CH₃ and C27–CH₃, between H27 and H28, and between H28 and H24 in oxolane **54** derived from **3** through a five-step process [(i) PPh₃, I₂, imidazole; (ii) Mg; (iii) DDQ, (iv) *m*-CPBA (yielding an almost single diastereomer); (v) CSA.] (Fig. 4).
34. Selected spectral data of **3**: A colorless oil; $[\alpha]_D^{26} -18.0$ (*c* 1.70, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 0.99 (3H, d, *J* = 6.8 Hz), 1.21 (3H, s), 1.53–1.60 (2H, m), 1.63–1.65 (1H, m), 1.69 (3H, d, *J* = 1.0 Hz), 1.71–1.94 (3H, m), 2.53–2.63 (1H, m), 3.60 (1H, dt, *J* = 5.8, 11.7 Hz), 3.66 (1H, ddd, *J* = 2.4, 5.8, 11.2 Hz), 3.81 (3H, s), 3.85 (2H, m), 3.97–4.03 (1H, m), 4.37 (2H, s), 5.28 (1H, dd, *J* = 1.3, 9.6 Hz), 6.87 (2H, d, *J* = 8.6 Hz), 7.25 (2H, d, *J* = 8.6 Hz); ¹³C NMR (75 Hz, CDCl₃, ¹³CDCl₃ as 77.0 ppm) δ 13.9 (CH₃), 22.6 (CH₃), 26.9 (CH₃), 27.5 (CH₂), 29.0 (CH), 37.9 (CH₂), 48.3 (CH₂), 55.2 (CH₃), 65.2 (CH₂), 71.0 (CH₂), 76.0 (CH₂), 78.7 (CH), 83.7 (C), 113.7 (CH × 2), 129.2 (CH × 2), 129.3 (C), 130.7 (C), 135.8 (CH), 159.1 (C); IR (film) ν_{\max} 3443, 2959, 2930, 2867, 1614, 1454, 1373, 1302, 1248, 1173, 1070, 1038, 821 cm⁻¹; HR-EIMS calcd for C₂₁H₃₂O₄ [M⁺]: 348.2301; found, 348.2300.